#### A Brief History of Molecular Dynamics

Or, an analysis of its place in high-performance computing

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# The Universe According to MD

 The Newtonian approximation of chemistry is a canonization of the way freshmen and sophomore students are taught to think of molecules.



 $+\sum_{i,j} \sum_{k \neq i} \frac{1}{q \neq j} / r \neq ij - 4\epsilon \neq ij (\sigma \neq ij / r \neq ij) + 4\epsilon \neq ij (\sigma \neq ij / r \neq ij) + 12 ]$ 

# Building on a remarkable success

- The Cornell charge set (parm94, Amber ff95) has been passed down through a long lineage: ff99, ff99SB, ff99SB-ILDN, ff14SB.
- Much faster computers enable QM calculations vastly more sophisticated than HF/6-31G\*.
- Hydrogen steric models have changed since 1994. A great deal more effort, in aggregate, has gone toward modifying torsions than any other aspect of the Amber force field.
- Electrostatics are perhaps the most physically meaningful parameters in the force field, and are easy to derive in an automated fashion.

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"If once you start down the dark path, forever will it dominate your destiny."

– Yoda

 Scheme (I) is pretty hard. Scheme (II) is only relevant in a vacuum. Scheme (III) is wrong.



• Scheme (I) is pretty hard. Scheme (II) is only relevant in a vacuum. Scheme (III) is wrong.



 Scheme (IV) is, again, wrong. (V) is, again, hard if performed with post Hartree-Fock methods.



 Scheme (IV) is, again, wrong. (V) is, again, hard if performed with post Hartree-Fock methods.



Energy of the polarized wave function Energy of the polarized wave function

## What viable alternative is left?

Back to scheme (II). It's incomplete, but it's straightforward and not yet wrong.



wave function

Energy of the unpolarized wave function

> Electostatic Interactions

Electostatic Interactions

Energy of the polarized wave function

# A new charge model: IPolQ

- The target potential is an average of two MP2 / cc-pvTZ calculations:
  - The molecular conformation in vacuum, and...
  - In a reaction field due to a bath of TIP4P-Ew water
- The fitting cycle is iterative



## A New Steric Repulsion set



 In order to bring hydration free energies of side chain analogs into agreement with experiment, Lennard-Jones parameters of five polar heavy atom types were adjusted.

## Two charge sets in one: Extended REsP



# Let the Model Drive

 A model may fit its training data and an independently generated test set. But what will it produce if allowed to drive energy minimizations?



 Any molecular model will make compromises against its benchmark, over- and under-estimating the energy in different regions. In actual simulations, the trend is always toward structures that are scored too favorably.

## Show the Model its Mistakes

 Sometimes the errors are very pronounced. In those cases, re-introducing the results of model-guided structure optimizations back into the training set produces a much more reliable model.



 In the case of Lysine dipeptide (above), new mistakes are evidently found in the second generation, but the model appears to be fixed by the third.

# ff14ipq: The first IPolQ protein model

First, let's look at the backbone behavior:



 The first generation of ff14ipq (solution phase charges paired with vacuum phase torsions) puts all of the minima in the right places.

# ff14ipq: The first IPolQ protein model

 Subsequent generations refine the Alanine dipeptide PMF:

Generation 1:

- 28,000 MP2 energies
- Amino acid
  dipeptides, Ψ
  tripeptides, and
  tetrapeptides
- Among the largest data sets as of early 2013



# ff14ipq: The first IPolQ protein model

 Subsequent generations refine the Alanine dipeptide PMF:

#### Generation 3:

- 65,000 MP2 energies
- Force-field
  optimized results of generations 1 and 2
- Artificially low minima were also eliminated in many amino acids



## Protein simulations and stability mature

 What appear to be interest excursions may just be incomplete models.



# Stability, but too much?

 β-sheets are too stable; in K19, lysine head groups contact the backbone too much.



# ff14ipq: What was the problem?

• The salt bridges are too stable: it seems that our polar atom Lennard Jones meddling did this.



 The over-stabilizations tend to be on the order of 1 kcal/mol, similar to an estimate in the ff14ipq paper.

# ff15ipq: The next IPolQ protein model

 Rather than make polar heavy atoms bigger, focus on the hydrogens. Nitrogens are central to the problems in hydration free energies.



f15ipq: Bigger polar hydrogens

 Introduce angle fitting alongside dihedrals. This helps relieve a lot of strain that would otherwise spill over into fitted torsion parameters.

## How effective is angle fitting?

|   | Ala(5) Scores |      |      |     | K19   | GB1                          |
|---|---------------|------|------|-----|---|------------------------------|
|   | Orig          | DFT1 | DFT2 | KLL | (α-helix)                                     | (β-hairpin)                  |
| ff14ipq   | 1.3           | 2.6  | 1.5  | 1.4 | Metastable at 277K                            | Completely stable<br>at 300K |
| Change polar H Lennard-Jones radius to 1.5Å, refit torsions               |               |      |      |     |   |                              |
| ff15ipq-05  | 1.5           | 2.5  | 1.5  | 1.5 | Unstable                                      | Completely stable            |
| Add N-CA-C, CA-C-N, and C-N-CA for neutral, Glycine, (+) and (-) residues |               |      |      |     |   |                              |
| ff15ipq-06  | 0.7           | 2.0  | 0.8  | 0.7 | Unstable                                      | Unstable                     |
| Decrease polar H Lennard-Jones radius to 1.3Å                             |               |      |      |     |   |                              |
| ff15ipq-08  | 0.7           | 2.3  | 1.0  | 0.7 | Too stable at 275K,<br>begins to melt at 315K | Unstable                     |
| Add CA-N-H, CA-C=O for neutral, Glycine, (+) and (-) residues             |               |      |      |     |   |                              |
| ff15ipq-09  | 0.6           | 2.7  | 1.0  | 0.7 | Metastable at<br>275-285K, 300-325K<br>t.b.d  | Melts at 300K                |

# How far does automation go?

 Automated parameter creation is bound to hit a wall: even if QM were perfect our ability to mimic it is not.



Difference in QM Methods, U(DFT) – U(MP2) Error, U(Tinker Toy) – U(MP2) -1: Favored relative to QM

## Electrostatic potential fitting

• The nuclear charges fit the quantum target with many compromises



# A force field is but a means to an end

• The fav8 peptide was engineered to study aromatic stacking between helicies in proteins. The peptide crystallizes with all solvent (water) accounted for in the unit cell.



 With the GPU-based pmemd code, this small system was simulated for nearly 10 μs (10 billion time steps).

# A force field is but a means to an end

 The simulated water density reproduces the natural electron density, not just the refined water positions



# MD as a Consumer of HPC

• Unmistakable choke points on the path between parameter development and biochemical simulation:



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